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PAPER

The influence of temperature on preparing tungsten doped vanadium dioxide films by sol-gel method

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10 October 2018Shuliang Dou^{1,3}, Weiyan Zhang^{2,3}, Yuemin Wang¹, Yi Wang¹, Xiang Zhang¹, Leipang Zhang¹, Lebin Wang¹, Jiupeng Zhao^{2,4} and Yao Li^{1,4} ¹ Centre for Composite Material and Structure, Harbin Institute of Technology, Harbin, People's Republic of China² School of Chemical Engineering and Technology, Harbin Institute of Technology, 150001, Harbin, People's Republic of China³ These authors contributed equally to this work.⁴ Authors to whom any correspondence should be addressed.E-mail: yaoli@hit.edu.cn and jpzhao@hit.edu.cn**Keywords:** vanadium dioxide, sol-gel method, optics, thermochromic film, smart windowSupplementary material for this article is available [online](#)**Abstract**

The influence of annealing temperature on the preparation of W- doped VO₂ films is investigated in the paper. W dopant is introduced by pouring the melt V₂O₅ into ammonium tungstate solution. And the W- doped VO₂ films are obtained via dip-coating method followed by annealing at argon atmosphere. The evolution of crystal structure, valence value of surface element, surface morphology, roughness and thermochromic performance are studied at different temperature. And different concentration of W doped VO₂ (M) films are prepared by the same method, which exhibit excellent performance. The proposed technique provides a path for fabricating W- doped VO₂(M) solar modulating coatings.

1. Introduction

Vanadium dioxide is a widespread interesting material since its novel metal-insulator transition (MIT), accompanied with distinct optical tunability in selective near-infrared [1]. It changes from metal state with opaque in near infrared regions to insulator with transparency at 68 °C [2]. The unique property has a promising application on smart device, such as microbolometers or field effect transistor [3], storage systems [4], sensors [5], electronic switching devices [6], and smart windows [7–9].

However, vanadium dioxide thin films have not been used in practical applications owing to their slightly higher transition temperature (T_t) compared to room-temperature (RT). Doping is an effective way to reduce the T_t . Particularly, the T_t is shifted to RT by doping films with high-valence ions, for instance, molybdenum [10] and tungsten [11]; as well as silicon [12] and fluorine [13]. Among these dopants, tungsten is found to be most effective, lowering T_t by 23 °C per 1 at% [14–16]; while 15 °C per 1 at% has also been reported for Mo [17].

However, the perfect stoichiometric VO₂ is difficult to fabricate because of the abundant vanadium oxides, such as V₃O₅, V₂O₅, [18] and so on. Besides, the main phases of VO₂ exist four forms, VO₂(A), VO₂(B) and VO₂(M/R), which have stable structures in similar conditions during VO₂ growth [19]. Even so, a wide range of methods are applied for VO₂ (M)-based film fabrication, such as sol-gel [20], magnetron sputtering deposition [21], wet chemical approaches [22, 23], chemical vapor deposition (CVD) [24], thermolysis method [25], and pulsed laser deposition (PLD) [26]. The sol-gel method is a low cost, facile coating on large scale surface, easy control of reaction kinetics and atomic doping and lower sintering temperature way to prepare VO₂ film [20, 27, 28]. In a typical sol-gel process for VO₂ synthesis, costly tetravalent alkoxide precursor, such as vanadium tetrabutoxide, was coated onto substrates, and then underwent high temperature annealing under inert atmosphere [25]. Up to now, researchers have developed the sol-gel methods for VO₂ thin film synthesis. For example, Gao *et al* developed an interesting VOCl₂ precursor containing PVP polymer [7, 8, 29–32]. Besides, inorganic sol-gel method is a facile method because its simple process and costless starting material [17, 33, 34].

Huang *et al* reported a involved mixing water with molten V_2O_5 powders to obtain $V_2O_5 \cdot nH_2O$ gels [35]. And the Mo^{5+} and W^{5+} ions were doped into VO_2 films by mixing ammonium molybdate and ammonium tungstate with V_2O_5 powder following by melting at $800^\circ C$, respectively.

Here, a facile method to prepare W-doped VO_2 films has been presented by modifying inorganic sol-gel method. And the W dopant was introduced by pouring the melt V_2O_5 into ammonium tungstate solution. And the influence of temperature is investigated in the annealing process from precursor to W-doped films. It seems that the W-doped VO_2 (M) film can be obtained at $600^\circ C$, which exhibits excellent performance on smart window. The results support an insight on the understanding of the preparation for W-doped VO_2 (M) film by inorganic sol-gel method.

2. Experiment

2.1. starting material

Vanadium pentoxide powder (V_2O_5 , AR) and Polyvinylpyrrolidone (PVP, K30, AR) were used as starting material to prepare sol, which were purchased from Tianjin Guangfu Fine Chemical Research Institute. Ammonium tungstate (AR) was used as W source. All the reagents were purchased and used without further purification, except for the ammonium tungstate was heated in vacuum oven at $200^\circ C$ to remove the adsorbed water.

2.2. Preparation of the precursor sol

W-doped precursor sol was fabricated by a water quenching method: 2.5 g V_2O_5 powder was melted at $850^\circ C$ for 10 min, and then poured into 100 ml ammonium tungstate solution (0.7 g L^{-1}) at room temperature (RT). The obtained slurry was stirred vigorously for 1 h, and diluted with de-ionized water (DI-18M Ω) water to 10 g L^{-1} . PVP (K30 6 wt%) was introduced to the sol to stabilize the sol at RT. The W-doped precursor sol was obtained after vigorous stirring for 10 h. the amount of doped W element was adjusted by the different concentration of ammonium tungstate solution. And the pure precursor was obtained by the same process, except for pouring melt V_2O_5 into DI water. All the samples used in this work were synthesized using these precursor sols, unless otherwise noted.

2.3. Fabrication of W-doped VO_2 films

Precursor films were deposited on fused-silica substrates ($1 \times 4\text{ cm}^2$) by dip coating method. After drying for 30 min at $80^\circ C$ to remove the excess solvent, the smooth precursor films were formed. These precursor films were then annealed under an argon atmosphere (purity of 99.999%), which was accompanied by crystallization and reduction from the V_2O_5 phase to the VO_2 phase. And the annealing temperature varied from $400^\circ C$ to $650^\circ C$ with an interval of $50^\circ C$, and the corresponding samples were signed as WV400, WV450, WV500, WV550, WV600 and WV650. The annealing time was chose as 5 h to ensure complete crystallization. The general fabrication process is illustrated in Scheme 1.

2.4. Measurements

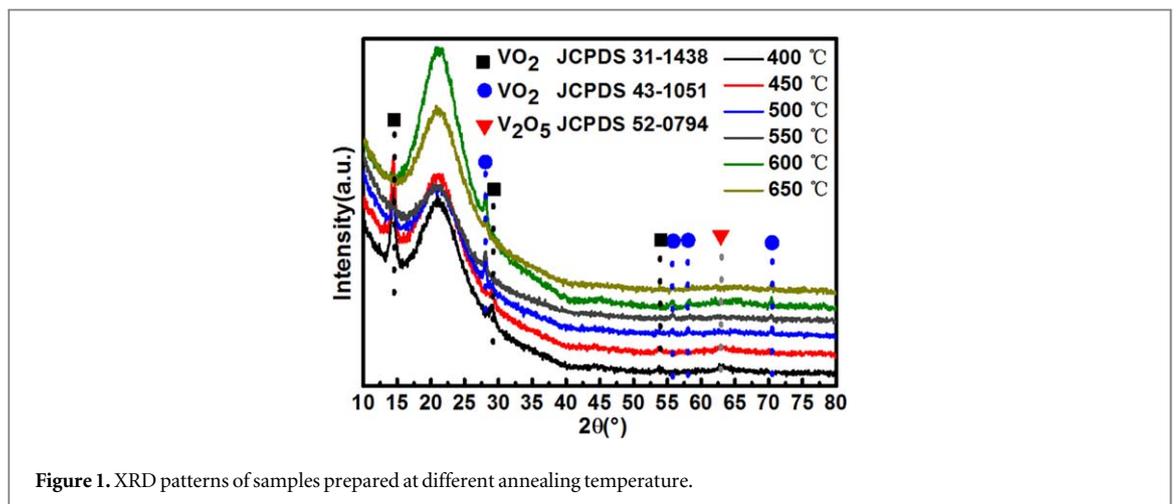
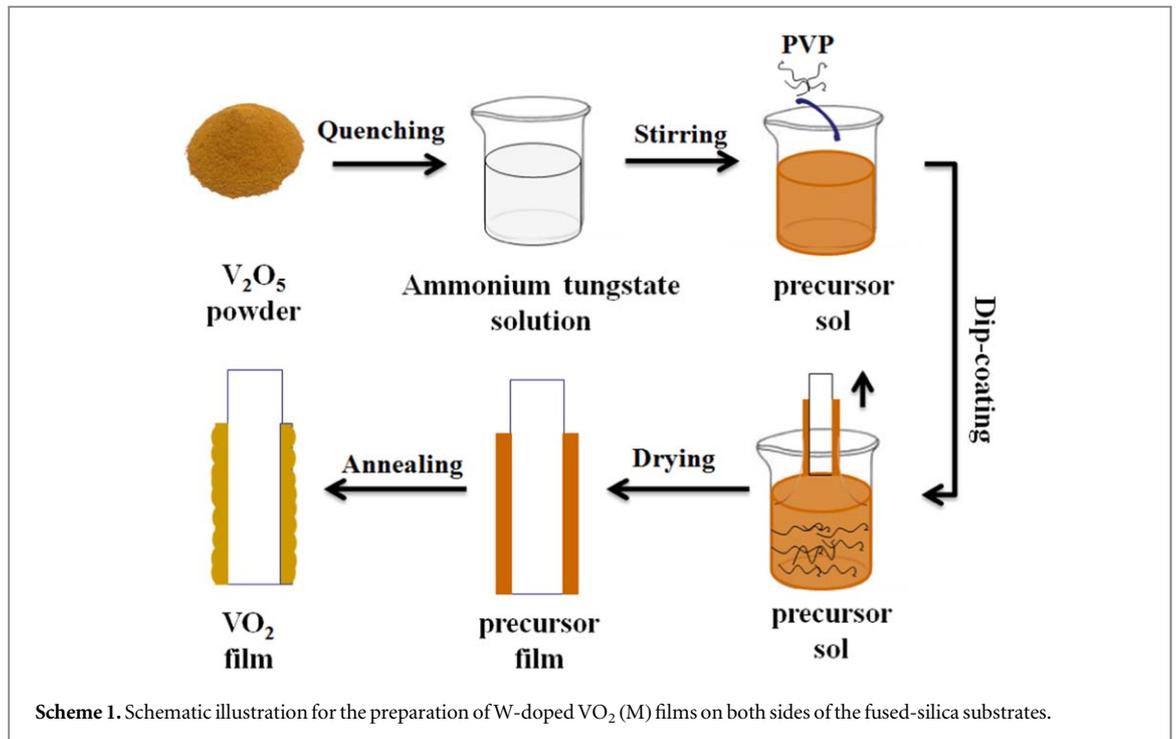
The crystalline phases of the films were determined using an x-ray diffraction (XRD) diffractometer (PANalytical B V, Model Xpert Pro, primary monochromatic Cu-K α radiation). The morphology and thickness of the films were determined by a field-emission scanning electron microscope (SUPRA 55 SAPPHIRE). The surface roughness was measured by an atomic force microscope (AFM, Dimension Icon, Bruker). x-ray photoelectron spectroscopy (XPS) studies were conducted with a PHI 5700 ESCA System using Al K α radiation (1486.6 eV).

The transmittance of the W-doped VO_2 films was measured using a UV-vis-NIR spectrophotometer (Lambda-950, Perkin Elmer) equipped with a film heating unit over the wavelength range of 250–2500 nm. Hysteresis loops were measured by collecting the transmittance of films at a fixed wavelength (2000 nm) at an approximate interval of $2.0^\circ C$ by using a fiber optic spectrometer (Ocean optics, NIRQuest 256–2.5). The temperature was measured with a thermocouple in contact with the film and was controlled by a temperature-controlling unit. The temperature errors were smaller than $0.5^\circ C$ based on repeated measurements.

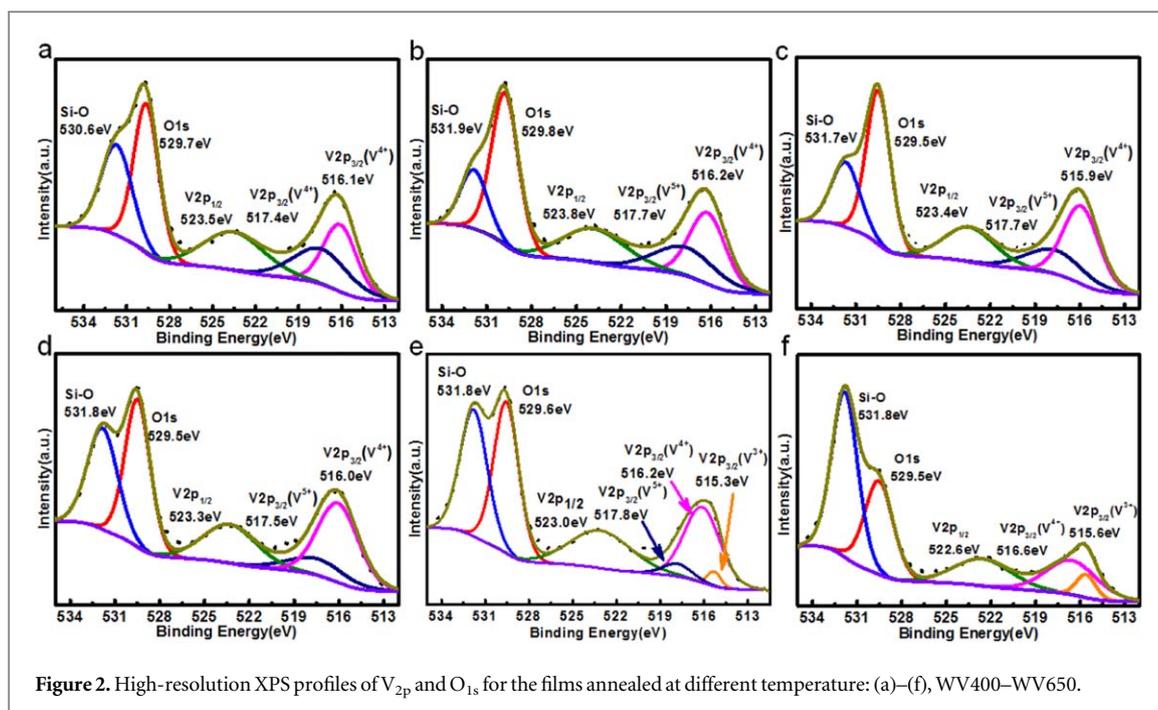
3. Results and discussion

3.1. Crystal structure

The phase evolution of the W-doped films was characterized by XRD. Figure 1 shows the XRD patterns for the films. And no phase of W based oxide is observed at patterns of all the samples. It is readily seen from the patterns that the sample WV400 is composed of two phases of VO_2 (B) (JCPDS 31–1438) and V_2O_5 (JCPDS 52–0794),



respectively [36–38]. And the main peaks of (011) and (002) crystal planes for B phase VO₂ are amplified in the figure S1 is available online at stacks.iop.org/MRX/6/016408/mmedia, which are in 14.4° and 29°, respectively. Besides, the diffraction peaks of the V₂O₅ phase is weak and the VO₂ (B) is strong, which means the main phase in the sample is VO₂ (B). When the annealing temperature increases to 450 °C, the secondary diffraction peak of VO₂(B) at 29° is weaker, while the main diffraction peak at 14.4° is still strong (figure S1). No monoclinic phase (JCPDS 43–1051) is observed in pattern of the sample WV450, and a tiny diffraction peak of V₂O₅ can be confirmed as V₂O₅ at 63.1°. VO₂ (M) appears in the pattern of sample WV500 [39–41], accompanying with a relative weak diffraction peak of VO₂ (B). And the predominant M phase exhibits a strong diffraction peak at 28°, which can be well assigned to the (011) crystal plane. However, although the intensity of diffraction peak for M phase is enhanced in the pattern of the sample WV550, a slight diffraction peak of VO₂(B) still exists at 14.4°. Furthermore, the intensity of diffraction peaks exhibit more stronger for M phase and no other phase is observed at the patterns of the sample WV600. However, as the annealing temperature increasing to 650 °C, the main diffraction peaks of W-doped VO₂ film disappears and only a weak peak survives as the (011) plane of M phase. Besides, the broad diffraction peaks at 15°–35° originate from the fused-silica substrates. It seems that higher annealing temperature may degenerate the crystallinity of M phase. Besides, the phase of precursor films change from V₂O₅ to VO₂ (M) via an intermediate phase of VO₂ (B). Furthermore, no coexist of three phases is observed in all the patterns, which is different to the phase transformation of pure VO₂ (M) film [42].



3.2. XPS analysis

The evolution of surface chemical composition and their valence states of the W-doped films annealed at different temperature are investigated by XPS. The valence states of V element are shown in figure 2. It seems that the valence states of V element in different samples vary obviously and comprise V^{5+} , V^{4+} and V^{3+} . For the sample WV400, as shown in figure 2(a), the deconvolution peak of $V_{2p_{3/2}}$ spectrum can be resolved into two components of 517.4 eV and 516.1 eV, which can be attributed to V^{5+} and V^{4+} , respectively [43–45]. As the annealing temperature increasing, the $V_{2p_{3/2}}$ spectra of WV450, WV500 and WV550 can be still deconvoluted into V^{5+} and V^{4+} , respectively (figure 2(b)), and the intensity of V^{4+} deconvolution peak enhances gradually. Furthermore, the V^{4+}/V ratio in the samples enlarged as the temperature increasing (table S1). Three valence states of V^{5+} , V^{4+} and V^{3+} coexists in the sample WV600, simultaneously. And it suggests that it is difficult to prepare pure M phase W-doped VO_2 film. And the V^{4+}/V ratio increases to largest of 83.8% (table S1). Furthermore, the V^{5+} valence state of W-doped VO_2 film disappears when the annealing temperature increased to 650 °C, accompanying with lower V^{4+}/V ratio.

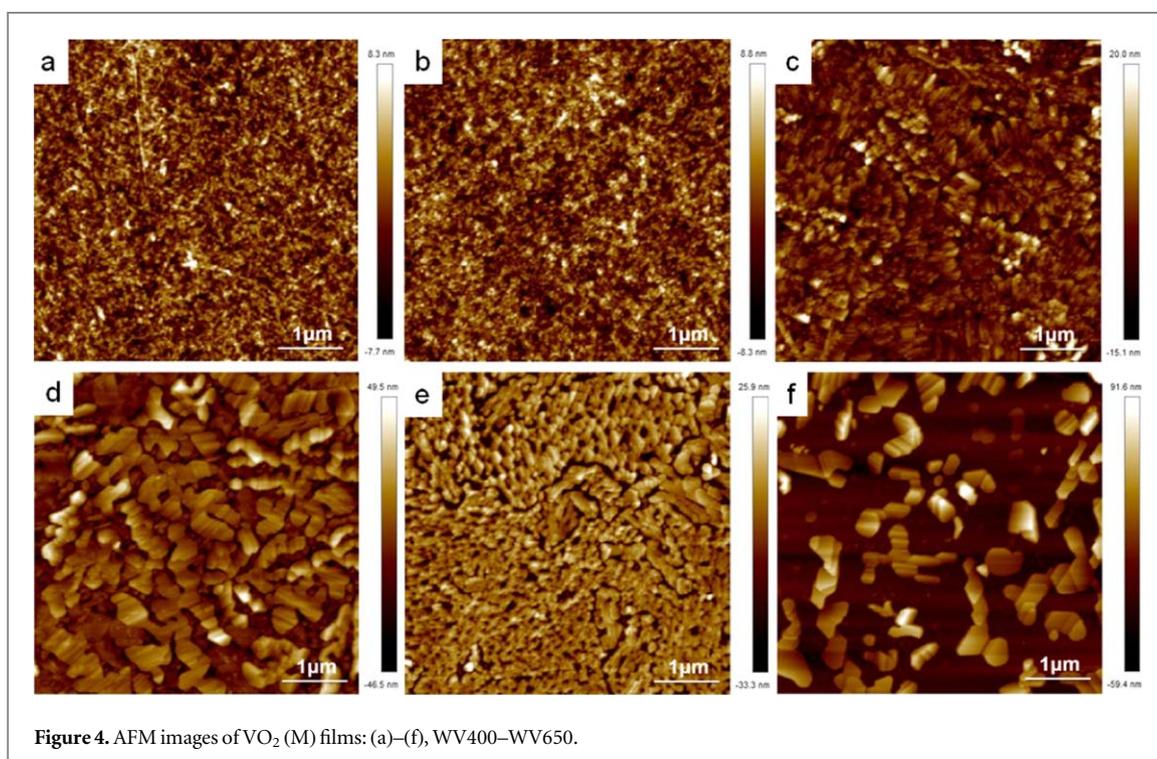
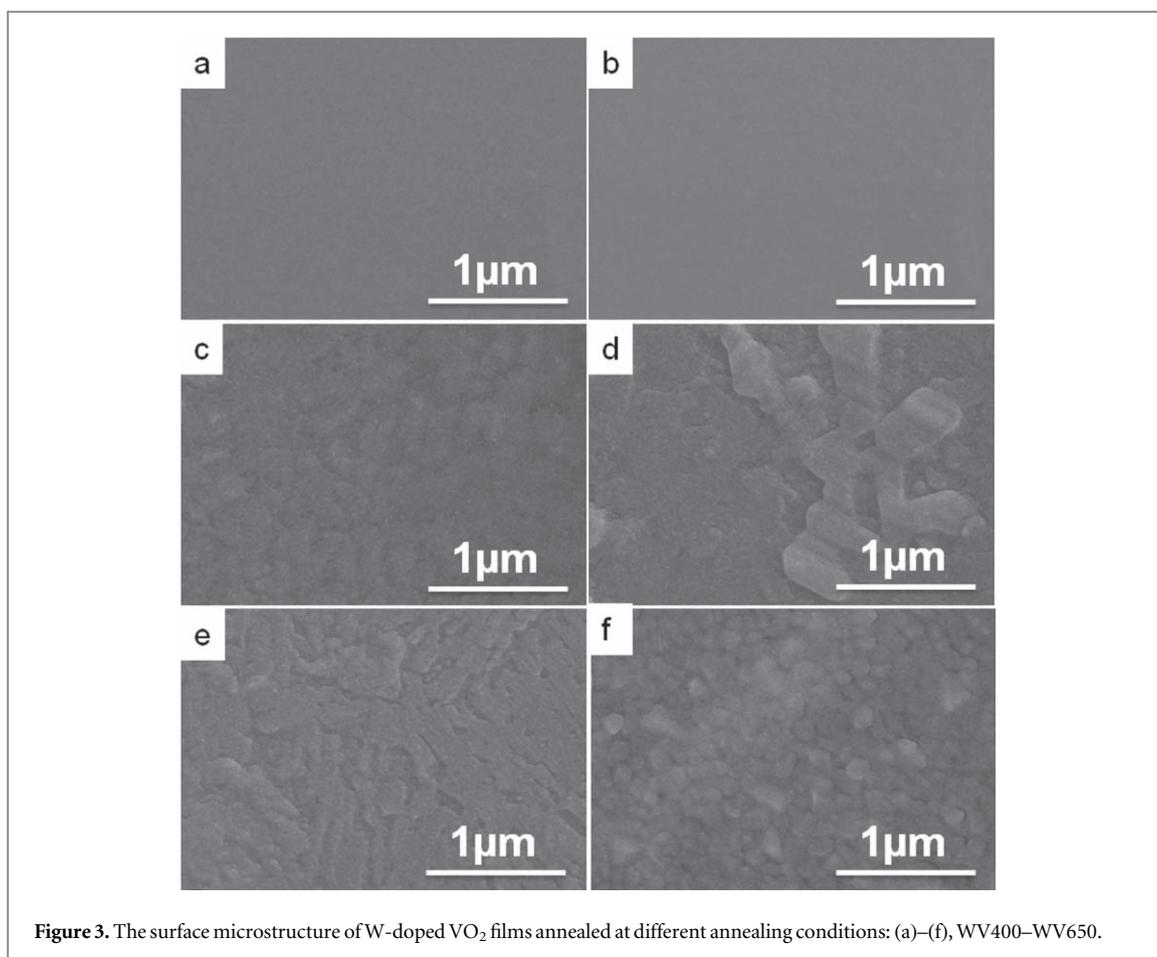
The corresponding deconvolution peak of W_{4f} spectra in the samples are also investigated. As shown in figure S2, the W element exists as W^{6+} in all the films [11, 20, 46], which means that the valence state of W element has no change throughout all the annealing temperature. Besides, the $W/(W + V)$ ratio varied randomly, which means the complicated phase transformation process from V_2O_5 to VO_2 .

Since the precursor sol comprises three typical N elements of $-C-N$, $W-N$ and NH_4^+ , the evolution of N is traced at different annealing temperature to understand the annealing process insightfully. As shown in figure S3, it seems that the deconvolution peaks in profiles of the samples degenerated as the annealing temperature increasing. And it is corresponding to the decomposition of PVP and NH_4^+ in the W-doped films. Besides, the existence of N_{1s} deconvolution peak suggests that N residual in the sample W650 (figure S3(f)), which means the uncompleted decomposing of PVP.

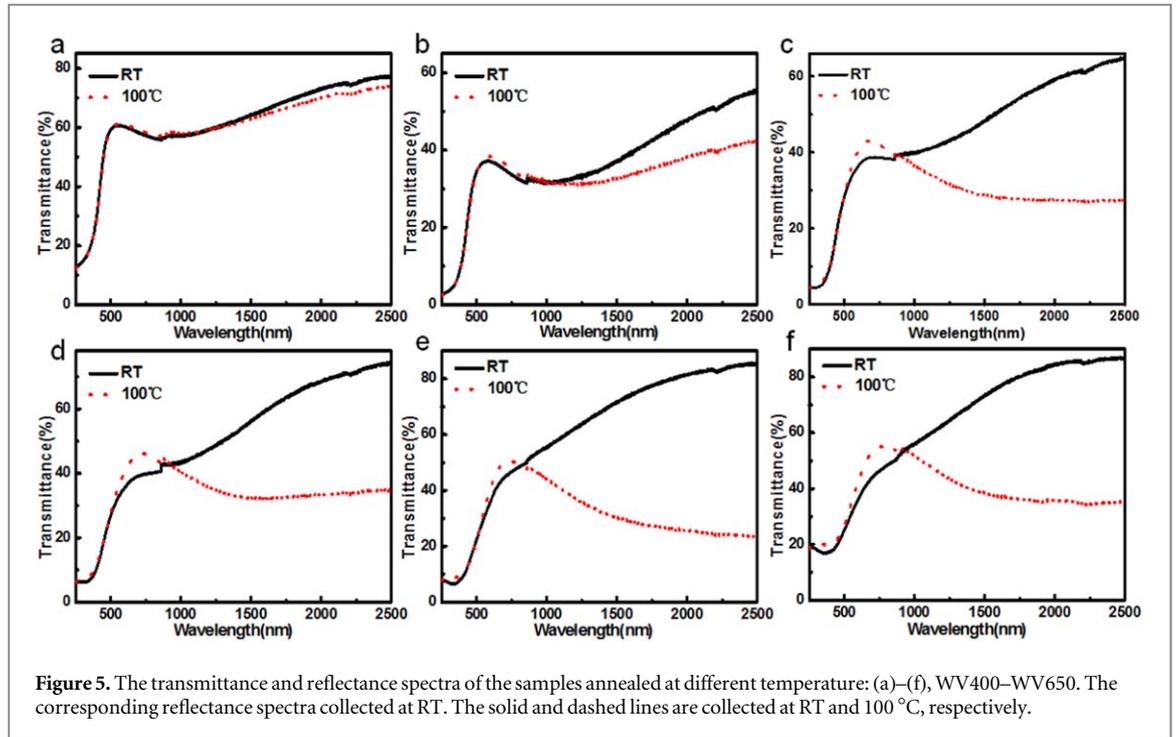
3.3. Surface morphology

The evolution of the surface morphology of the films was investigated by SEM. As shown in figure 3, it suggests that the sample WV400 and WV450 are smooth and dense (figures 3(a) and (b)), and the films WV500, WV550, WV600 and WV650 are composed of micro-structure (figures 3(c), (d) and (e)). However, it seems that the sample WV500 is comprised of dense submicro particles. And the sample WV550 consists of inhomogeneous component in the surface. Furthermore, the surface of sample WV600 is ice-crystal like microstructure, which is similar to the undoped VO_2 film annealed at 500 °C. Besides, the film WV650 consists of dense particles on the surface, and the grain diameter is about one hundred nanometer. It seems that the change on the film surface texture with increasing annealing temperature can be attributed to the effects of surface tension and the complex decomposing process of precursor gel film.

The AFM measurement is used to investigate the variation of microstructure in height of the samples, as shown in figure 4. The samples exhibit the same evolutions as the SEM results. Furthermore, the average



roughness values (R_a , R_q) and maximum roughness values (R_{max}) of the samples are shown in figure S4(a). The roughness of the samples increases with the annealing temperature, which means the distinct morphology changes in the annealing process. And the R_a and R_q changes from several nanometers (WV400) to the finally resulting of dozen nanometers in height (WV650). Besides, the R_{max} is overwhelmed the R_a and R_q , which means



the uneven surface of the samples. Interestingly, the roughness of sample WV600 lower those value of both WV550 and WV650. It can be seen from the AFM images that the WV600 is composed of more smaller micro-structure, which lead to lower roughness.

3.4. Thermochromic performances

The transmittance spectra of the W-doped films annealed at different temperature are investigated at RT and 100 °C, which corresponds to the two states of the VO₂ (M) film. As shown in figure 5, the variation of transmittance for different samples is significant difference at near infrared wavelength. The variation of transmittance at 2000 nm between RT and 100 °C is defined as $\Delta T_{2000\text{nm}}$, which is obtained by equation (1):

$$\Delta T_{2000\text{nm}} = T_{2000\text{nm}}(\text{M}) - T_{2000\text{nm}}(\text{R}) \quad (1)$$

Where the $T_{2000\text{nm}}(\text{M}/\text{R})$ is the transmittance at 2000 nm measured at RT and 100 °C, respectively.

The $\Delta T_{2000\text{nm}}$ of the samples is gradually enlarged as the annealing temperature increasing to 600 °C and slightly lowered at 650 °C (figure. S3(b)). And it seems that the $\Delta T_{2000\text{nm}}$ changes with the component of VO₂(M) phase in the films. The $\Delta T_{2000\text{nm}}$ of samples W1 and W2 is very small, which is attributed to the rarely VO₂ (M) phase in the films. And the samples W3 and W4 shows larger $\Delta T_{2000\text{nm}}$, where the VO₂(M) is main phase in the films. Furthermore, the WV600 exhibits the largest $\Delta T_{2000\text{nm}}$ in all the samples, which is in accordance with the dominated VO₂(M) phase and the largest V⁴⁺/V ratio in the film. Besides, the $\Delta T_{2000\text{nm}}$ of sample WV650 degenerates slightly but still larger than those of WV400, WV450, WV500 and WV550. It seems that the ‘particle’ surface of sample W6 enhance the transmittance at 100 °C, which lead to smaller $\Delta T_{2000\text{nm}}$ compared with W5.

Typically, the main optical performances of VO₂ thermochromic smart windows, including luminous transmittance (T_{lum} , 390–780 nm) and solar modulate ability (ΔT_{sol} , 250–2500 nm), are obtained using the following equations:

$$T_{\text{lum}/\text{sol}} = \int \varphi_{\text{lum}/\text{sol}}(\lambda) T(\lambda) d\lambda / \int \varphi_{\text{lum}/\text{sol}}(\lambda) d\lambda \quad (2)$$

$$\Delta T_{\text{sol}} = T_{\text{sol}}(30\text{ }^\circ\text{C}) - T_{\text{sol}}(100\text{ }^\circ\text{C}) \quad (3)$$

where $T(\lambda)$ is the recorded film transmittance, φ_{lum} is the standard luminous efficiency function for the photopic vision of human eyes, and φ_{sol} is the solar irradiance spectrum for air mass 1.5 (corresponding to the Sun at 37° above the horizon).

The T_{lum} and ΔT_{sol} of the samples are calculated by equations (2) and (3). As shown in figure S3(b), the ΔT_{sol} shows the same evolution as the $\Delta T_{2000\text{nm}}$ in different samples, while the T_{lum} is opposite. And the T_{lum} changes slightly in the samples W4, W5 and W6. Especially, the sample WV600 exhibits overwhelming ΔT_{sol} and the smallest T_{lum} compared to others, which may be the result of the coupling interaction between crystal structure and surface morphology.

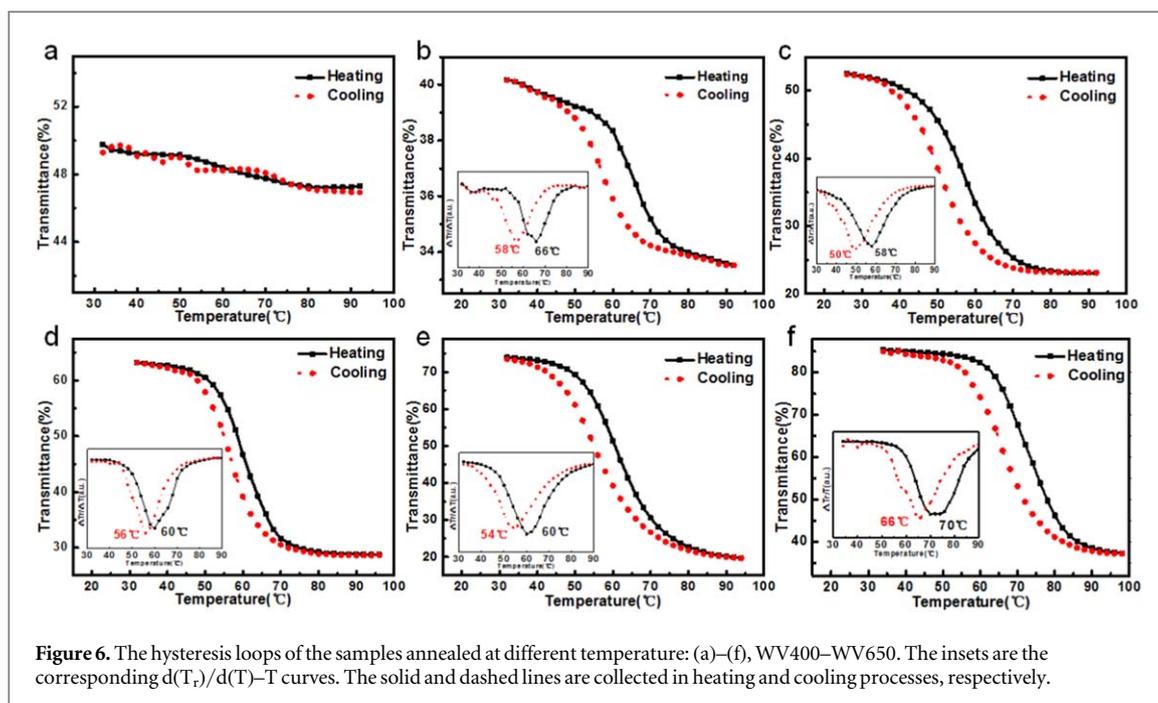


Figure 6. The hysteresis loops of the samples annealed at different temperature: (a)–(f), WV400–WV650. The insets are the corresponding $d(T_r)/d(T)$ – T curves. The solid and dashed lines are collected in heating and cooling processes, respectively.

Table 1. The phase transition temperature and hysteresis loop width of the samples.

Samples	T_h	T_c	T_t	ΔT
W400	—	—	—	—
W450	66	58	62	8
W500	58	50	54	8
W550	60	54	57	6
W600	60	54	57	6
W650	70	64	67	6

The hysteresis loops of the films were measured by recording the transmittance variation in heating and cooling processes at 2000 nm, as shown in figure 6. It can be seen that the Sample WV400 exhibits no hysteresis loops. And the other samples show obvious hysteresis loops, which means the existing $\text{VO}_2(\text{M})$ in the films.

The $d(T_r)/d(T)$ – T curves were used to describe the definition of the transition parameters of heating and cooling branches. The temperature corresponding to the peak is defined as the phase transition temperature of the heating and cooling branches, denoted as T_h and T_c , respectively. For cooling branches with the appearance of two peaks in the $d(T_r)/d(T)$ – T curves, the T_c values are determined by the main peak. The phase transition temperature (T_t) of the film is defined as average value of T_h and T_c . The hysteresis loop width (ΔT) is defined as the difference between T_h and T_c . The phase composition, the optical performances, transition temperature and hysteresis loop width of the films are summarized in table 1. The T_t of all the samples is lower than the pure VO_2 film of 68 °C. And the ΔT is more narrow than that of pure VO_2 film. And the sample WV600 shows T_t of 57 °C and ΔT of 6 °C.

4. Conclusions

In conclusion, the influence of annealing temperature on preparation of W-doped VO_2 films by inorganic sol method is investigated. The phase changes from V_2O_5 to $\text{VO}_2(\text{M})$ via an intermediate phase of $\text{VO}_2(\text{B})$ without coexistence of three phases. Besides, the valence of V element exists three valence state in the W-doped $\text{VO}_2(\text{M})$ films. Furthermore, the N element in the PVP develops to the N residual on the VO_2 surface. And the surface roughness of the samples enlarges as the annealed temperature increasing, except for a slight reduce at 600 °C. The ratio of W and V element changes at the annealing process, which suggest the complex formation of W-doped VO_2 film. Besides, the optimized annealing temperature is 600 °C accompanied with largest ΔT_{sol} and pure M phase VO_2 . And the phase transition temperature is lowered by the introduced W dopant. The proposed technique gives an insight understanding for preparing W-doped $\text{VO}_2(\text{M})$ solar modulating coatings.

Acknowledgments

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ORCID iDs

Yao Li  <https://orcid.org/0000-0003-3859-8135>

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